# A Practical Route to C-8 Substituted Fluoroquinolones

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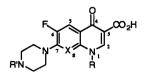
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Abstract: The ortho metalation of N-(tert-butoxycarbonyl)-3,4-difluoroaniline, with t-BuLi at -78°C in THF occured regioselectively at C-2 position. The resulting lithiated species reacted with a variety of electrophiles to give 2-substituted-3,4-difluoroanilines in good yields after hydrolysis. These compounds have been readily transformed into C-8 substituted fluoroquinolones. The nucleophilic substitution at C-7 by piperazines and the alkylation at N-1 on these quinolones have also been performed.

#### Introduction

In recent years, many molecular modifications of the parent structures of quinolones have been carried out in order to develop agents with higher potency and broader bacterial coverage<sup>1</sup>. In fact, almost all clinically useful quinolones bear a fluorine atom at the C-6 position and a nitrogen heterocycle, usually a piperazine ring, at the C-7 position. These antibacterial agents, generally known as fluoroquinolones, are now a major class of antiinfectives with significant potential for continued development<sup>2</sup>. Notable examples are Norfloxacin, Pefloxacin, Enoxacin, Ciprofloxacin, Amifloxacin, Ofloxacin and Fleroxacin (Figure 1).

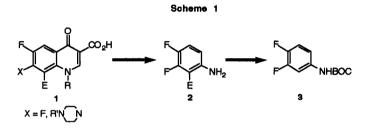


	х	R	R'
Norfloxacin	сн	CH <sub>2</sub> CH <sub>3</sub>	н
Perfloxacin	сн	CH <sub>2</sub> CH <sub>3</sub>	СН3
Enoxacin	N	CH <sub>2</sub> CH <sub>3</sub>	н
Ciprofloxacin	сн	c-C <sub>3</sub> H <sub>5</sub>	н
Amifloxacin	сн	NHCH3	СН <sub>3</sub>
Fleroxacin	CF	CH <sub>2</sub> CH <sub>2</sub> F	Сн₃
Ofloxacin	Ŷ	↓сн₃	СН₃

Figure 1. Clinically significant quinolone antibacterials

In order to perform wider structure-activity studies, the development of general synthetic methods for the preparation of differently substituted quinolones is highly desirable. We wish to report herein a general method of synthesis of fluoroquinolones substituted at the C-8 position from commercially available 3,4difluoroaniline. This starting material is currently used for the preparation of some important C-8 unsubstituted quinolones such as Pefloxacin and Norfloxacin.

Scheme 1 briefly shows our retrosynthetic plan. It was expected that C-8 substituted quinolones 1 would be easily prepared from 2-substituted-3,4-difluoroanilines 2 according to the Gould-Jacobs reaction sequence<sup>3</sup> for the construction of the quinolone ring. This sequence involves condensation with diethyl ethoxymethylenemalonate and intramolecular Friedel-Crafts acylation as the main steps. An attractive general route to the synthesis of the key intermediates 2 would be the regioselective C-2 ortho metalation<sup>4</sup> of the N-BOC derivative of 3,4-difluoroaniline (3) and its further reaction with electrophiles. This approach would be based on the cooperative metalation effects of the strong ortho directing NHBOC group<sup>5</sup> and the weaker ortho directing ability of  $F^{6}$ .



**Results and Discussion** 

The required compound, N-(*tert*-butoxycarbonyl)-3,4-difluoroaniline (3), was readily prepared by reaction of 3,4-difluroaniline with di-*tert*-butyl dicarbonate in refluxing THF (97% yield). When a solution of 3 in THF was treated with 2.2 equiv. of *t*-BuLi at -78°C for 1 h a pale yellow solution was formed that after quenching with an excess of D<sub>2</sub>O gave regioselectively and quantitatively the deuterated compound **5a** (Table 1, entry a). This result shows that under these conditions<sup>7</sup> a complete ortho deprotonation occurs at C-2 with subsequent formation of the metalated anion 4. This carbanion also reacted, under very mild conditions, with a wide variety of electrophiles, such as 1,2-dibromoethane<sup>8</sup> (entry b), methyl disulfide (entry c), MeI (entry d), DMF (entry e), acetaldehyde (entry f) and cyclohexanone (entry g) to give the expected products **5**. Excellent yields were obtained after chromatography (78-94%). In all cases the reaction was highly regioselective (the C-6 substituted isomers were not detected by <sup>1</sup>H NMR). On the other hand, when dianion **4** was quenched with acetaldehyde or cyclohexanone and the reaction mixture was left at room temperature overnight, cyclization of the primary product (**5f** or **5g**) occurred spontaneously to give the benzoxazinone derivative **6** in good yield (Scheme 2).

With C-2 substituted carbamates 5 in hand we undertook their transformation into C-8 substituted quinolones 1. The overall sequence from compounds 5c, 5d and 5h (the latter obtained by reduction of 5e with NaBH4) is depicted in Scheme 3. Acid deprotection of BOC-derivatives 5 (HCl, Et<sub>2</sub>O, reflux) afforded the substituted fluoroanilines 2 (80-87% yield) whose condensation with diethyl ethoxymethylenemalonate, at

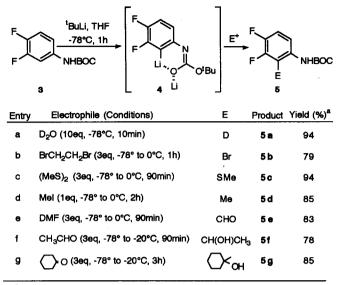
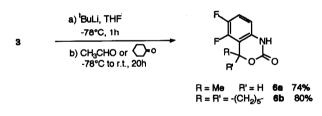


Table 1: Regioselective preparation of N-BOC-2-substituted-3,4-difluoroanilines 5

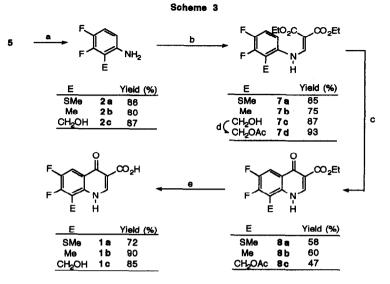
<sup>a</sup> Yield in pure isolated product after chromatography.

Scheme 2



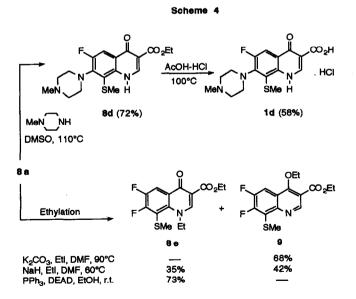
110°C, gave enamines 7 in good yields (75-87%). Intramolecular acylation of 7a, 7b and 7d (the alcohol 7c was protected as its acetate 7d) by heating at 220°C in diphenyl ether<sup>3</sup> led to the formation of ethyl 4oxoquinoline-3-carboxylic esters 8 (47-60% yield) which were hydrolyzed under basic conditions (NaOH, 100°C) to the corresponding free carboxylic acids (quinolones 1, 72-90% yield).

Finally, we attempted nucleophilic substitution at C-7 and alkylation at N-1 in order to achieve the usual pattern of substitution of quinolone antibacterials (Figure 1). However, the nucleophilic substitution of fluorine at C-7 position by different piperazines<sup>9</sup> was not successful on quinolones 1. Fortunately, satisfactory results were obtained when these reactions were carried out starting from their ethyl esters precursors 8 (Scheme 4). Thus, the reaction of 8a with N-methyl piperazine in DMSO at 110°C yielded exclusively the C-7 substituted compound 8d (72%), which was hydrolyzed in acid conditions (HCl, AcOH, 100°C) to afford the 7-piperazinyl quinolone 1d (58%). Regarding the alkylation at N-1 position we observed a competition between N- and O-ethylation in quinolone 8a, which was very dependent on the nature of the base and the ethylating agent. For instance, whereas ethylation of 8a with K<sub>2</sub>CO<sub>3</sub>/EtI in DMF<sup>10</sup> at 90°C for 5 days gave exclusively



(a) HCl (c), Et<sub>2</sub>O, r.t.; (b) EtOCH≃C(CO<sub>2</sub>Et)<sub>2</sub>, 110°C; (c) Ph<sub>2</sub>O, 220°C; (d) Ac<sub>2</sub>O, Py, r.t.; (e) NaOH, EtOH-H<sub>2</sub>O, 100°C.

the O-ethylated quinoline 9 in 68% yield (20% of unreacted 8a was recovered), its ethylation with PPh<sub>3</sub>/DEAD/EtOH in CH<sub>2</sub>Cl<sub>2</sub> at r.t. gave selectively the N-ethylated quinolone 8e in good yield (73% after purification). An intermediate situation was observed when 8a was ethylated with NaH/EtI in DMF, under these conditions 35% of 8e and 42% of 9 were obtained after chromatography.



In summary, the ortho metalation of N-BOC-3,4-difluoroaniline with t-BuLi in THF at -78°C is completely regioselective. The C-2 substituted N-BOC anilines 5, obtained after reaction with electrophiles, can be readily applied to the synthesis of a wide range of C-8 substituted fluoroquinolones.

#### **Experimental Section**

Melting points are uncorrected. IR spectra were recorded as thin films from CHCl<sub>3</sub> or nujol (only strong absortions are reported. <sup>1</sup>H NMR spectra were taken at 200 MHz, on a Bruker WP-200-SY, in the solvents indicated. High resolution mass spectra were recorded under e.i. at 70 eV. All reagents or solvents were used as comercial grades, except THF which was distilled from Na/benzophenone and DMF which was distilled from CaH<sub>2</sub>. Flash column chromatography was performed on silica gel Merck-60 (230-430 mesh).

**N-(***tert*-**butoxycarbonyl)-3,4-difluoroaniline** (3). A solution of 1.30 g (10.1 mmol) of 3,4difluoroaniline and 2.42 g (11.1 mmol) of di-*tert*-butyldicarbonate in 10 ml of dry THF was heated at 60°C for 15 h. The solvent was removed in vacuo and the residue was dissolved in 15 ml of ethyl acetate. The solution was washed successively with 1 % HCl (2 x 10 ml) and 5 % NaHCO<sub>3</sub> (2 x 10 ml). The organic layer was dried (MgSO<sub>4</sub>) and evaporated in vacuo. The residue was recrystallized from hexane to give 2.24 g (97%) of carbamate **3**, m.p.135-137°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 7.43 (ddd, 1H, J = 12.3, 7.5 and 2.6 Hz, ArH), 7.05 (m, 1H, ArH), 6.90 (m, 1H, ArH), 6.48 (broad s, 1H, NH), 1.52 (s, 9H, <sup>t</sup>Bu); IR (nujol) 3300, 2880, 2800, 1690, 1610, 1430, 1390, 1370, 1290, 1170; MS *m/e* 229 (M<sup>+</sup>, 10), 173 (37), 129 (43), 101 (13), 86 (18), 84 (22), 59 (20), 57 (100); Anal. Calcd. for C<sub>11</sub>H<sub>13</sub>F<sub>2</sub>NO<sub>2</sub>: C, 57.64; H, 5.67; N, 6.11. Found: C, 57.63; H, 5.56; N, 6.06.

General procedure for dilithiation of 3 and further reaction with electrophiles. A 1.6 M solution of *tert*-butyllithium in hexane (22.5 ml, 33.0 mmol) was added slowly to a solution of 3 (3.43 g, 15.0 mmol) in dry THF (40 ml) maintained at -78°C under argon. After 1 h at -78°C the appropriate electrophilic reagent was added and the solution was allowed to warm to the temperature for the reaction time indicated in Table 1. Then, saturated NH<sub>4</sub>Cl (15 ml) was added and the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 15 ml). The combined organic layers were dried (MgSO<sub>4</sub>) and evaporated to give crude products **5a-g** which were purified by flash chromatography.

**2-Deutero-3,4-difluoro-N**-(*tert*-butoxycarbonyl)aniline (5a). 3.24 g, 94 % (eluent: hexane/ethyl acetate 6/1), m.p.124-126°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 7.05 (m, 1H, ArH), 6.9 (ddd, 1H, J = 8.6, 4.3 and 1.4 Hz, ArH), 6.48 (broad s, 1H, NH), 1.51 (s, 9H, 'Bu); IR (nujol) 3300, 2880, 2800, 1690, 1610, 1430, 1390, 1370, 1290, 1170; Anal. Calcd. for C<sub>11</sub>H<sub>12</sub>DF<sub>2</sub>NO<sub>2</sub>: C, 57.39; H, 6.09; N, 6.09. Found: C, 57.42; N, 5.65; N, 6.27.

**2-Bromo-3,4-difluoro-N-**(*tert*-butoxycarbonyl)aniline (5b). 3.06 g, 66% (eluent: hexane/ethyl acetate 30/1), m.p.56-57°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 7.29 (ddd, 1H, J = 9.4, 4.6 and 2.3 Hz, ArH), 7.12 (m, 1H, ArH), 6.88 (broad s, 1H, NH), 1.53 (s, 9H, 'Bu); IR (nujol) 3340, 2950, 2840, 1745, 1700, 1590, 1510, 1400, 1370, 1270, 1150; Anal. Calcd. for C<sub>11</sub>H<sub>12</sub>BrF<sub>2</sub>NO<sub>2</sub>: C, 42.85; H, 4.05; N, 4.51. Found: C, 42.86; H, 4.22; N, 4.54.

**3,4-Difluoro-2-methylthio-N-(***tert***-butoxycarbonyl)aniline** (5c). 3.88 g, 94% (eluent: hexane/ethyl acetate 25/1), m.p.53-54°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 7.95 (ddd, 1H, J = 9.2, 4.6 and 1.9 Hz, ArH), 7.75 (broad s, 1H, NH), 7.12 (m, 1H, ArH), 2.36 (s 3H, SCH<sub>3</sub>) 1.53 (s, 9H, <sup>t</sup>Bu); IR (nujol) 3360, 2910, 2840, 1730,

1700, 1580, 1500, 1450, 1390, 1270, 1220, 1160; MS m/e 275 (M<sup>+</sup>, 32), 202 (30), 175 (81), 160 (60), 57 (100); HRMS Calcd. for C<sub>12</sub>H<sub>15</sub>F<sub>2</sub>NO<sub>2</sub>S 275.0791. Found 275.0793.

**3,4-Difluoro-2-methyl-N**-(*tert*-butoxycarbonyl)aniline (5d). 3.10 g, 85% (eluent: hexane/ethyl acetate 20/1), m.p.97-98°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 7.30 (m, 1H, ArH), 6.96 (m, 1H, ArH), 6.19 (broad s, 1H, NH), 2.19 (d, 3H, J = 2.6 Hz, CH3), 1.51 (s, 9H, <sup>1</sup>Bu); IR (nujol) 3340, 2945, 2850, 1650, 1510, 1500, 1440, 1370, 1290, 1250, 1150; Anal. Calcd. for C<sub>12</sub>H<sub>15</sub>F<sub>2</sub>NO<sub>2</sub>: C, 59.26; H, 6.17; N, 5.76. Found: C, 59.42; H, 6.04; N, 5.74.

**2,3-Difluoro-6**-(*tert*-butoxycarbamoyl)benzaldehyde (5e). 3.20 g, 83% (eluent: hexane/ethyl acetate 20/1), m.p.130-131°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 10.42 (broad s, 1H, NH), 10.39 (s, 1H, CHO), 8.25 (ddd, 1H, J = 9.5, 4.2 and 2.1 Hz, ArH), 7.40 (m, 1H, ArH), 1.55 (s, 9H, <sup>4</sup>Bu); IR (nujol) 3285, 2940, 2840, 1730, 1670, 1660, 1595, 1370, 1300, 1240, 1210, 1170; MS *m/e* 257 (M<sup>+</sup>, 8), 201 (18), 184 (12), 157 (29), 129 (30), 86 (34), 84 (57), 69 (11), 57 (100); Anal. Calcd. for C<sub>12</sub>H<sub>13</sub>F<sub>2</sub>NO<sub>3</sub>: C, 56.03; H, 5.58; N, 5.45. Found: C, 55.89; H, 5.86; N, 5.47.

**3,4-Difluoro-2-(1-hydroxyethyl)-N-(***tert***-butoxycarbonyl)aniline** (**5f**). 3.19 g, 78% (eluent: hexane/CH<sub>2</sub>Cl<sub>2</sub> 2/1), m.p.114-115°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 8.47( broad s, 1H, NH), 7.71 (ddd, 1H, J = 9.2, 4.1 and 1.6 Hz, ArH), 6.96 (m, 1H, ArH), 5.47 (q, 1H, J = 6.7 Hz, CHOH), 2.49 (broad s, 1H, OH), 1.55 (d, 3H, J = 6.7 Hz, CH<sub>3</sub>) and 1.51 (s, 9H, <sup>1</sup>Bu); IR (nujol) 3410, 3310, 2930, 2840, 1690, 1600, 1530, 1415, 1360, 1270, 1160; MS m/e 327 (M<sup>+</sup>, 16), 271 (65), 253 (86), 236 (21), 227(34), 209 (100), 180 (36), 166 (80); HRMS Calcd. for C<sub>13H17</sub>F<sub>2</sub>NO<sub>3</sub> 273.1176. Found 273.1172.

**3,4-Difluoro-2-(1-hydroxycyclohexyl)-N-(***tert***-butoxycarbonyl)aniline** (**5g**). 4.17 g, 85% (eluent: hexane/CH<sub>2</sub>Cl<sub>2</sub> 2/1), m.p.156-157°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 9.51 (broad s, 1H, NH), 7.82 (ddd, 1H, J = 9.3, 5.1 and 2.3 Hz, ArH), 7.00 (m, 1H, ArH), 2.30-1.25 (m, 10H, cyclohexyl), 1.59 (s, 9H, <sup>1</sup>Bu); IR (nujol) 3405, 3240, 2920, 2840, 1675, 1595, 1500, 1400, 1360, 1270, 1160; MS *m/e* 273 (M<sup>+</sup>, 22), 217 (68), 199 (12), 182 (45), 173 (40), 155 (100), 127 (19), 69 (16), 57 (40); HRMS Calcd. for C<sub>17</sub>H<sub>23</sub>F<sub>2</sub>NO<sub>3</sub> 327.1646. Found 327.1649.

General procedure for the preparation of trisubstituted anilines 2. Concentrated HCl (20 ml) was added to a solution of the corresponding carbamate 5 (4.17 mmol) in Et<sub>2</sub>O (42 ml) and the resulting mixture was vigorously stirred at room temperature for 40 min.-17h. 2M NaOH was added until pH>10 and the mixture was extracted with Et<sub>2</sub>O (3 x 15 ml). The combined organic layers were washed with H<sub>2</sub>O (2 x 10 ml), dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated. As soon as the crude anilines 2 were purified by chromatography (hexane/ethyl acetate 1/2 as eluent) they were used in the following reaction due to their limited stability.

**3,4-Difluoro-2-methylthioaniline** (**2a**). 628 mg, 86% (reaction time: 17 h); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 6.93 (m, 1H, ArH), 6.42 (ddd, 1H, J = 9.1, 4.1 and 2.0 Hz, ArH), 4.29 (broad s, 2H, NH<sub>2</sub>), 2.32 (s, 3H, SCH<sub>3</sub>). **3,4-Difluro-2-methylaniline** (**2b**). 476 mg, 80% (reaction time: 40 min); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 6.80 (m, 1H, ArH), 6.35 (m, 1H, ArH), 3.55 (broad s, 2H, NH<sub>2</sub>), 2.10 (d, 3H, J = 2.1 Hz, CH<sub>3</sub>).

**3,4-Difluoro-2-hydroxymethylaniline** (2c). 577 mg, 87% (reaction time: 40 min); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 6.92 (m, 1H, ArH), 6.38 (ddd, 1H, J = 8.4, 3.8 and 2.1 Hz, ArH) 4.75 (d, 2H, J = 2.0 Hz, CH<sub>2</sub>OH) 3.5-2.5 (broad s, 3H, NH<sub>2</sub> and OH).

General procedure for the preparation of methylenemalonates 7. A mixture of diethyl ethoxymethylenemalonate (10.64 ml, 3.18 mmol) and aniline 2 (3.18 mmol) was heated at 110°C. After 2 h the mixture was cooled to room temperature and hexane (10 ml) was added. The resulting precipitate was filtered

off, washed with hexane, and dried. This crude malonate 7 was used in the following reaction without further purification.

**Diethyl (3,4-difluoro-2-methylthioanilinyl)methylenemalonate (7a)**. 0.93 g, 85% (reaction time: 2.30 h), m.p.88-89°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 11.54 (d, 1H, J = 13.7 Hz, NH), 8.47 (d, 1H, J = 13.7 Hz, HC=C), 7.20 (m, 1H, ArH), 7.01 (ddd, 1H, J = 9.3, 4.1 and 1.8 Hz, ArH), 4.35 (q, 2H, J = 7.1 Hz, CH<sub>2</sub>), 4.26 (q, 2H, J = 7.1 Hz, CH<sub>2</sub>), 2.43 (s, 3H, SCH<sub>3</sub>), 1.39 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>), 1.34 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>); IR (nujol) 2910, 2840, 1665, 1630, 1580, 1440, 1390, 1285, 1250, 970, 880, 790; Anal. Calcd. for C<sub>15</sub>H<sub>17</sub>F<sub>2</sub>NO<sub>4</sub>S: C, 52.03; H, 4.92; N, 4.11. Found: C, 52.17; H, 4.93; N, 4.05.

**Diethyl (3,4-difluoro-2-methylanilinyl)methylenemalonate** (7b). 0.75 g, 75% (reaction time: 1.30 h), m.p. 103-104°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 11.10 (d, 1H, J =13.1 Hz, NH), 8.41 (d, 1H, J = 13.1 Hz, HC=C), 7.07 (m, 1H, ArH), 6.94 (ddd, 1H, J = 9.1, 4.2 and 1.6 Hz ArH), 4.32 (q, 2H, J = 7.1Hz, CH<sub>2</sub>), 4.25 (q, 2H, J = 7.1 HZ, CH<sub>2</sub>), 2.31 (d, 2H, J = 2.2 Hz, CH<sub>3</sub>), 1.39 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>), 1.33 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>); IR (nujol) 2910, 2840, 1670, 1630, 1600, 1460, 1380, 1250, 1090, 1020, 1000, 790; Anal. Calcd. for C<sub>15</sub>H<sub>17</sub>F<sub>2</sub>NO<sub>4</sub>: C, 57.61; H, 5.44; N, 4.38. Found C, 57.51; H, 5.43; N, 4.47.

**Diethyl (3,4-difluoro-2-hydroxymethylanilinyl)methylenemalonate (7c).** 0.91 g, 87% (reaction time: 1 h), m.p.133-134°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 11.49 (d, 1H, J= 13.5 Hz, NH), 8.40 (d, 1H, J = 13.5 Hz, HC=C), 7.18 (m, 1H, ArH), 7.01 (ddd, 1H, J = 9.0, 3.9 and 1.8 Hz, ArH), 4.87 (d, 2H, J = 2.1 Hz, CH<sub>2</sub>OH), 4.31 (q, 2H, J = 7.0 Hz, CH<sub>2</sub>), 4.25 (q, 2H, J = 7.0 Hz, CH<sub>2</sub>), 2.85 (broad s, 1H, OH), 1.37 (t, 3H, J = 7.0 Hz, CH<sub>3</sub>), 1.33 (t, 3H, J = 7.0 Hz, CH<sub>3</sub>); IR (nujol) 3360, 2910, 2840, 1670, 1630, 1580, 1450, 1400, 1285, 1250, 1110, 970, 790; Anal. Calcd. for C<sub>15</sub>H<sub>17</sub>F<sub>2</sub>NO<sub>5</sub>: C, 54.63; H, 5.16; N, 4.12. Found: C, 54.71; H, 5.17; N, 4.25.

**Diethyl (2-acetoxymethyl-3,4-difluoroanilinyl)methylenemalonate** (7d). To a solution of 1.23 g (3.75 mmol) of 7c in 1.25 ml (15.00 mmol) of pyridine was added at room temperature 0.35 ml (3.75 mmol) of acetic anhydride. The resulting mixture was stirred for 90 minutes, then a solution of 10% HCl was added until pH 2-3. The mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 5 ml), the combined organic layers were washed with 5% NaHCO<sub>3</sub> (2 x 5 ml), dried (MgSO<sub>4</sub>) and evaporated to afford 1.29 g (93%) of 7d, m.p.107-108°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 11.32 (d, 1H, J = 13.5 Hz, NH), 8.35 (d, 1H, J = 13.5 Hz, HC=C), 7.23 (m, 1H, ArH), 6.98 (ddd, 1H, J = 9.1, 4.1 and 1.9 Hz, ArH), 5.20 (d, 2H, J = 2.0 Hz, CH<sub>2</sub>OAc), 4.31 (q, 2H, J = 7.0 Hz, CH<sub>2</sub>), 4.25 (q, 2H, J = 7.0 Hz, CH<sub>2</sub>), 2.20 (s, 3H, CH<sub>3</sub>CO), 1.37 (t, 3H, J = 7.0 Hz, CH<sub>3</sub>) 1.33 (t, 3H, J = 7.0 Hz, CH<sub>3</sub>); IR (CHCl<sub>3</sub>) 2910, 1680, 1640, 1600, 1490, 1370, 1220, 1180, 1110, 795; Anal. Calcd. for C<sub>17</sub>H<sub>19</sub>F<sub>2</sub>NO<sub>5</sub>: C, 54.99; H, 5.12; N, 3.77. Found: C, 54.96; H, 5.35; N, 3.41.

General procedure for the preparation of quinolone ethyl esters 8. The methylenemalonate 7 (3.00 mmol) was added to diphenyl ether (6 ml) and the mixture was heated at 220°C for 1-10 h. The solution was cooled to room temperature and the resulting precipitate was filtered off, washed with hexane, and dried. The crude ester 8 was used in the following reaction without further purification.

**Ethyl 6,7-difluoro-1,4-dihydro-8-methylthio-4-oxoquinoline-3-carboxylate (8a).** 520 mg, 58% (reaction time: 10 h), m.p.267-270°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 9.46 (s, 1H, HC=C), 8.45 (t, 1H, J = 8.2 Hz, ArH), 4.74 (q, 2H, J = 8.2 Hz, CH<sub>2</sub>), 2.64 (s, 3H, SCH<sub>3</sub>), 1.57 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>); IR (nujol) 2920, 2840, 1680, 1650, 1590, 1450, 1370, 1260, 1230; Anal. Calcd. for C<sub>13</sub>H<sub>11</sub>F<sub>2</sub>NO<sub>3</sub>S: C, 52.17; H,3.68; N, 4.68. Found: C, 51.81; H, 3.67; N, 4.33.

Ethyl 6,7-difluoro-1,4-dihydro-8-methyl-4-oxoquinoline-3-carboxylate (8b). 481 mg, 60% (reaction time: 6 h), m.p.289-291°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 9.34 (s, 1H, HC=C), 8.28 (t, 1H, J = 8.3 Hz, ArH), 4.69 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>), 2.75 (d, 2H, J = 2.3 Hz, CH<sub>3</sub>), 1.53 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>); IR (nujol) 2920, 2850, 1700, 1630, 1590, 1480, 1460, 1380, 1300, 1190, 1110; Anal. Calcd. for C<sub>13</sub>H<sub>11</sub>F<sub>2</sub>NO<sub>3</sub>: C, 58.43; H, 4.12; N, 5.24. Found: C, 57.93; H, 4.10; N, 5.12.

**Ethyl 8-acetoxymethyl-6,7-difluoro-1,4-dihydro-4-oxoquinoline-3-carboxylate** (8c). 458 mg, 47% (reaction time: 1 h), m.p.245-250°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 9.13 (s, 1H, HC=C), 8.12 (t, 1H, J = 8.7 Hz, ArH), 5.82 (d, 2H, J = 2.0 Hz, CH<sub>2</sub>OAc), 4.51 (q, 2H, J = 7.0 Hz, CH<sub>2</sub>), 2.11 (s, 3H, CH<sub>3</sub>CO<sub>2</sub>), 1.47 (t, 3H, J = 7.0 Hz, CH<sub>3</sub>).

Ethyl 6-fluoro-7-(4-methyl-1-piperazinyl)-1,4-dihydro-8-methylthio-4-oxoquinoline-3carboxylate (8d). 0.28 ml (2.57 mmol) of N-methylpiperazine were added to a solution of 0.31 g (1.03 mmol) of ester 8a in 12 ml of DMSO. The resulting solution was heated at 110°C for 13 h.The solvent was removed by distillation in vacuo (0.1 mm Hg) and the residue was treated with Et<sub>2</sub>O (10 ml). The precipitate was filtered off and purified by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub> / MeOH 7/1) to give 0.28 g (72%) of 8d, m.p.250-251°C (dec.); <sup>1</sup>H NMR (CF<sub>3</sub>CO<sub>2</sub>D)  $\delta$ : 9.32 (s, 1H, HC=C), 8.27 (d, 1H, J = 11.4 Hz, ArH), 4.68 (q, 2H, J = 7.2 Hz, CH<sub>2</sub>O), 4.1-3.3 (m, 8H, NCH<sub>2</sub>), 3.15 (s, 3H, NCH<sub>3</sub>), 2.39 (s, 3H, SCH<sub>3</sub>), 1.53 (t, 3H, J = 7.2 Hz, CH<sub>3</sub>); IR (nujol) 2910, 2840, 1690, 1600, 1580, 1530, 1450.

General procedure for the preparation of quinolones 1. A mixture of ester 8 (2.50 mmol), EtOH (22ml) and 2M NaOH (4.5 ml) was refluxed for 3-17 h. The solution was cooled to room temperature and acidified with 1% HCl. The resulting precipitate was filtered off, washed with H<sub>2</sub>O and dried. The crude quinolone 1 was purified by recrystallization.

**6,7-Difluoro-1,4-dihydro-8-methylthio-4-oxoquinoline-3-carboxylic acid** (1a). 0.49 g, 72% (reaction time : 17 h), m.p.289-291°C (recrystallized from DMF); <sup>1</sup>H NMR (CF<sub>3</sub>CO<sub>2</sub>D)  $\delta$ : 9.51(s, 1H, HC=C), 8.41(t, 1H, J = 8.7 Hz, ArH), 2.62 (s, 3H, SCH<sub>3</sub>); IR (nujol) 3100, 2910, 2840, 1700, 1625, 1460, 1370, 1220, 1100, 910, 800; Anal. Calcd. for C<sub>11</sub>H<sub>7</sub>NO<sub>3</sub>S: C, 48.71; H, 2.58; N, 5.17. Found C, 48.32; H, 2.49; N,4.96.

**6,7-Difluoro-1,4-dihydro-8-methyl-4-oxoquinoline-3-carboxylic** acid (1b). 0.53 g, 90% (reaction time : 6.3 h), m.p.>300°C (dec.) (recrystallized from DMF); <sup>1</sup>H NMR (CF<sub>3</sub>CO<sub>2</sub>D)  $\delta$ : 9.44 (s, 1H, HC=C), 8.32 (t, 1H, J = 8.7 Hz, ArH), 2.79 (s, 3H, CH<sub>3</sub>); IR (nujol) 3100, 3060, 2910, 2840, 1700, 1620, 1570, 1470, 1380, 1360, 1220, 1005, 985, 800; Anal. Calcd. for C<sub>11</sub>H<sub>7</sub>F<sub>2</sub>NO<sub>3</sub>: C, 55.23; H, 2.93; N, 5.86. Found C, 54.98; H, 2.66; N, 5.75.

**6,7-Difluoro-8-hydroxymethyl-1,4-dihydro-4-oxoquinoline-3-carboxylic** acid (1c). 0.54 g, 85% (reaction time: 3 h), m.p.267-270°C (recrystallized from Toluene/ DMF 20/1); <sup>1</sup>H NMR (CF<sub>3</sub>CO<sub>2</sub>D)  $\delta$ : 9.60 (s, 1H, HC=C), 8.57 (t, 1H, J = 8.7 Hz, ArH), 5.70 (s, 2H, CH<sub>2</sub>OH); IR (nujol) 3200, 3100, 3040, 2910, 2840, 1685, 1620, 1465, 1370, 1235, 1000, 1040, 1010, 970; Anal. Calcd. for C<sub>11</sub>H<sub>7</sub>F<sub>2</sub>NO<sub>4</sub>: C, 51.76; H, 2.74; N, 5.49. Found: C, 52.01; H, 2.82; N, 4.92.

6-Fluoro-7-(4-methyl-1-piperazinyl)-1,4-dihydro-8-methylthio-4-oxoquinoline-3-carboxylic acid hydrochloride (1d). 0.38 g (1 mmol) of ester 8d was dissolved in 30 ml of a (1:4) mixture of acetic acid and concentrated HCl. The solution was refluxed at 110°C for 18 h and after cooling to room temperature, the precipitate was filtered off, washed successively with H<sub>2</sub>O and Et<sub>2</sub>O and dried. The crude acid was recrystallized from DMF:toluene to yield 0.20 g (58%) of 1d, m.p.289-290°C; <sup>1</sup>H NMR (CF<sub>3</sub>CO<sub>2</sub>D)  $\delta$ : 9.32 (s, 1H, HC=C), 8.28 (d, 1H, J = 11.5 Hz, ArH), 4.1-3.3 (m, 8H, NCH<sub>2</sub>), 3.16 (s, 3H, NCH<sub>3</sub>), 2.42 (s, 3H, SCH<sub>3</sub>);

## Ethylation of quinolone 8a.

<u>Method A:</u> To a solution of 229 mg (1 mmol) of **8a** in 4 ml of dry DMF was added 60 mg (1.5 mmol) of 60% NaH in oil. The solution was stirred for 15 min at room temperature. Then, 0.41 ml (5mmol) of EtI was added and the solution was heated at 60 °C for 3 h. H<sub>2</sub>O (5 ml) was added and the mixture was extracted with Et<sub>2</sub>O ( $3 \times 10 \text{ ml}$ ). The combined organic layers were dried and evaporated. The crude product was purified by flash chromatography (Cl<sub>2</sub>CH<sub>2</sub>/hexane 1/1) to give 114 mg (35%) of N-ethylated quinolone **8e** and 137 mg (42%) of O-ethylated quinoline **9**.

<u>Method B:</u> A solution of  $60 \ \mu l \ (0.41 \text{ mmol})$  of DEAD in 0.5 ml of dry THF was added dropwise to a solution of 100 mg (0.33 mmol) of quinolone ester **8a** and 0.1 g (0.41 mmol) of triphenylphosphine in 1 ml of dry THF. The solution was stirred overnight at room temperature. The solvent was removed in vacuo and Et<sub>2</sub>O was added to the residue to precipitate triphenylphosphine oxide and diethyl hydrazinedicarboxylate which were filtered off. The filtrate was evaporated and purified by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>) to give 79 mg (73%) of **9**.

Ethyl 6,7-difluoro-1,4-dihydro-N-ethyl-8-methylthio-4-oxoquinoline-3-carboxylate (8e). m.p.44-46°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 9.23 (s, 1H, HC=C), 7.91 (t, 1H, J = 8.5 Hz, ArH), 4.46 (q, 2H, J = 7.1 Hz, CH<sub>2</sub>), 4.33 (q, 2H, J = 7.1 Hz, CH<sub>2</sub>), 2.69 (d, 3H, J = 2.4 Hz, SCH<sub>3</sub>), 1.53 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>), 1.44 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>); IR (CHCl<sub>3</sub>) 2960, 1700, 1580, 1480, 1460, 1420, 1250, 1220, 1180, 1080, 1010, 880; Anal. Calcd. for C<sub>15</sub>H<sub>15</sub>F<sub>2</sub>NO<sub>3</sub>S: C, 55.09; H, 4.59; N, 4.28. Found: C, 54.83; H, 4.59; N, 4.04. **6,7-Difluoro-3-ethoxycarbonyl-4-ethoxy-8-methylthioquinoline** (9). m.p.113-115°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 8.46 (s, 1H, ArH), 8.36 (t, 1H, J = 8.7 Hz, ArH), 4.87 (q, 2H, J = 7.1 Hz, CH<sub>2</sub>), 4.41 (q, 2H, J = 7.1 Hz, CH<sub>2</sub>), 2.48 (s, 3H, SCH<sub>3</sub>), 1.47 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>), 1.42 (t, 3H, J = 7.1 Hz, CH<sub>3</sub>); IR (CHCl<sub>3</sub>) 2980, 1715, 1680, 1620, 1600, 1435, 1305, 1175, 1075, 870; MS *m/e* 327 (M<sup>+</sup>, 25), 282 (14), 255 (100), 212 (9), 182 (4), 151 (8), 113 (5); Anal. Calcd. for C<sub>15</sub>H<sub>15</sub>F<sub>2</sub>NO<sub>3</sub>S: C, 55.09; H, 4.59; N, 4.28. Found: C, 55.40; H, 4.73; N, 3.97.

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### **References and Notes**

1.- For some reviews about structure-activity studies see: a) Chu, D.T.; Fernandes, P.B. Antimicrob. Agents Chemother. 1989, 33, 131. b) Matsumoto, J.I.; Nishimura, Y.; Minamida, A. Chem. Pharm. Bull. 1988, 36, 1223. c) Klopman, G.; Macina, O.; Levinson, M.E.; Rosenkranz, H.S. Antimicrob. Agents Chemother. 1987, 31, 1831. d) Fujita, T. The role of QSAR in drug design: fact or fantasy?, Jolles, G.; Wooldridge, K.R.H., Eds; Academic Press: New York, 1984. For some recent publications see: e) Remuzon, P.; Bouzard, D.; Cesare P.D.; Essiz, M.; Jacquet, J.P.; Kiechel, J.R.; Ledoussal, B.; Kessler R.E.; Fung-Tomc J. J. Med. Chem. 1991, 34, 29. f) Ohta, M.; Koga, H. J. Med. Chem.. 1991, 34, 131. g) Chu, D.T.; Nordeen, C.W.; Hardy, D.J.; Swanson, R.N.; Giardina, W.J.; Pernet, A.G.; Plattner, J.J. J. Med. Chem.. 1991, 34, 168. Kiely, J.S.; Hutt, M.P.; Culbertson, T.P.; Bucsh, R.A.; Worth, D.F.; Lesheski, L.E.; Gogliotti, R.D.; Sesnie, J.C.; Solomon, M.; Mich, T.F. J. Med. Chem.. 1991, 34, 656.

2.- a) Wentland, M.P. in *The New Generation of Quinolones;* Siporin, C., Heifetz, C.L., Domagala, J.M., Eds., Marcel Dekker: New York, **1990**. b) Andriole, V.T. *The Quinolones*, Academic Press: New York, **1988**.

3.- a) Hayakawa, I.; Hiramitsu, T.; Tanaka, Y. Chem. Pharm. Bull. 1984, 32, 4907. b) Koga, H.; Itoh, A.; Murayama, S.; Suzue, S.; Irikura, T. J. Med. Chem. 1980, 23, 1358. c) Albrecht R., Development of antibacterial agents of the nalidixic acid type Prog. Drug. Res. 1977, 21, 9-104.

4.- For a recent review on directed ortho metalation see: Snieckus, V. Chem. Rev. 1990, 90, 879.

5.- a) Cho, I.S.; Gong L.; Muchowski, J.M. J. Org. Chem. 1991, 56, 7288. b) Clark, R.D.; Caroon, J.M. J. Org. Chem. 1982, 47, 2804. c) Muchowski, J.M.; Venuti M.C. J. Org. Chem. 1980, 45, 4798. d) Fuhrer, W.; Gschwend, H.W. J. Org. Chem. 1979, 44, 1133.

6.- a) Bridges, A.J.; Patt, W.C.; Stickney, T.M. J. Org. Chem. 1990, 55, 773. b) Gschwend, H.W.; Hamdan, A. J. Org. Chem. 1982, 47, 3652.

7.- It should be noted that deprotonation of 3 must be performed at very low temperature (-78°C) in order to avoid the elimination of the initially formed metalated species 4 into its corresponding benzyne.

8.- a) Wittig, G.; Harborth, G. Chem. Ber. 1944, 77, 306. b) Glaze, W.H.; Selman, C.M.; Ball, A.L.; Bray, L.E. J. Org. Chem. 1969, 34, 641.

9.- No substitution at all was detected when quinolones 1 were treated with N-methyl piperazine in DMSO at 110-150°C. Under these conditions a slow decarboxylation reaction was observed.

10.- By contrast C-8 unsubstituted quinolones gave exclusively the N-ethylated product after alkylation with K<sub>2</sub>CO<sub>3</sub>/IEt in DMF. See for instance ref. 3b and Domagala, J.M.; Hanna, L.D.; Heifetz, C.L.; Hutt, M.P.; Mich, T.F.; Sanchez, J.P.; Solomon, M. J. Med. Chem. **1986**, 29, 394.